

## Errata: *PRINCIPLES OF STABLE ISOTOPE DISTRIBUTION*

### Ch 1

- p.11 “These atoms have all been present since the Big Bang. ....” should read:  
“These atoms have all been present since their generation in stellar interiors and novae. ....”
- p. 13 line 5 from bottom: Change “5, 8, 147, and 149, ” to read ““5, 8, and 147”
- p. 14 “the Big Bang” should read “nucleosynthesis.”
- p. 31 eq. 1.20c change  $^{13}\text{C}/^{13}\text{C}$  to  $^{13}\text{C}/^{12}\text{C}$  in two places
- p. 36 prob 8: Change “0.1 g/l, ” to read “0.3 g/l”
- p. 37 problem 14, 10 keV should read 10 kV
- p. 37 prob 18: Change “how many ions per second are being transmitted. .... ” to read “how many molecules per second are being transmitted. .... , and what is their velocity if the pressure at the entrance of the capillary is 30 torr.”

### Ch 2

- p. 41. Here and throughout the book, the word “isotopomer” should be replaced by “isotopologue” to conform to IUPAC (1997).
- p. 42, line 7 from bottom: change “(equation 2.2a)” to read “(equation 2.2b)”  
also, in eq. 2.5c, R should be the gas constant **R**
- p. 43 2<sup>nd</sup> PP lines 7-8 equation 2.2b should read “equation 2.6
- p. 76 in line 7 from bottom:  $^{18}\text{O}^{16}\text{O}$  should read  $^{18}\text{O}/^{16}\text{O}$
- p. 84  
prob 5: change “confirm their calculated abundances and confirm”  
to read “confirm their calculated abundances with equation 2.8c and confirm”  
prob 6: Delete “or 2.8 c.”  
prob 7 : equation 2.21 should read equation 2.64 “or 2.8 c.”

### Ch 3

p. 91 bottom this definition of the “deuterium excess” differs from normal convention

p. 92 eq. 3.3b  $2.0 \pm 1 \text{ ‰}$  should read  $-2.0 \pm 1 \text{ ‰}$

p. 97. line 10 from bottom, change “typically thousands of grams per kilogram,” to read, “typically thousands of milligrams per kilogram”

p. 127 line 7: Change “may shorter or longer” to read “may be shorter or longer”

p.133 problem 1. For water J, the salinity should be 1.95 percent, not 19.5 per cent.

### Ch 4

p. 145 in eq. 4.16 and in the following line,  $R$  should be the gas constant  $R$

p. 167. Equation 4.49b change the right hand term from  $V \nabla \cdot C$  to read  $V \cdot \nabla C$ . Also, in line 3 from bottom, change “where  $\nabla \cdot C$  is the divergence of the concentration” to read, “where  $\nabla C$  is the concentration gradient.”

### Ch 5

p. 186  $+2$  to  $+16$  should read  $-2$  to  $+16$

p. 217 “...the average temperature of interaction of about  $350^\circ\text{C}$ ...” should read “...the average temperature of interaction of about  $250^\circ\text{C}$ ...”

### Appendix:

p. 240 In the lhs of equation A.3.10.4,  $v_{mm}^2$  should read  $v_{Mm}^2$

p. 241 eqn. A.3.12.4 For more accuracy in Stirling’s approximation, add the term  $0.5 \ln(2\pi x)$  to the terms,  $x \ln x - x$